



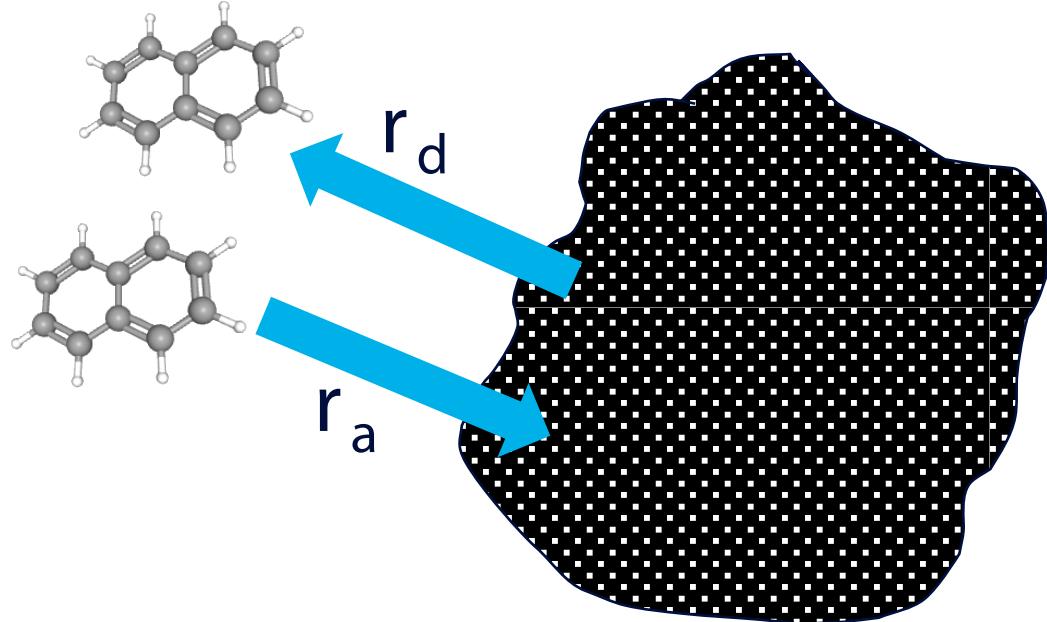
# Evaluation of Activated Carbon for Environmental Remediation Applications

Andy Harris, P.E., Senior Applications Engineer  
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# Outline

1. Adsorption equilibrium
2. Adsorption kinetics
3. Significance of kinetics on cap designs
4. Demonstration test: impact of carbon particle size on adsorption

# Adsorption Equilibrium and Kinetics



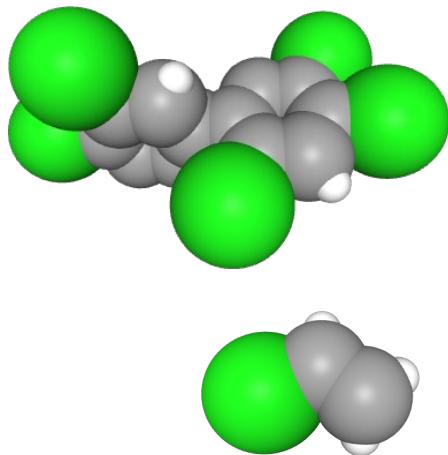
Equilibrium Loading Capacity → Carbon Consumption  
Kinetics

Transient (unsteady state):  $r_a \neq r_d$   
Equilibrium (steady state):  $r_a = r_d$

# Factors Affecting Equilibrium Loading Capacity

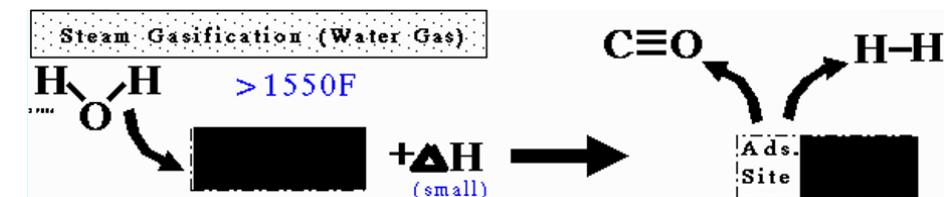
## Adsorbate

- Concentration
- Solubility
- Molar Volume
- Polarizability
- (Background Organics)



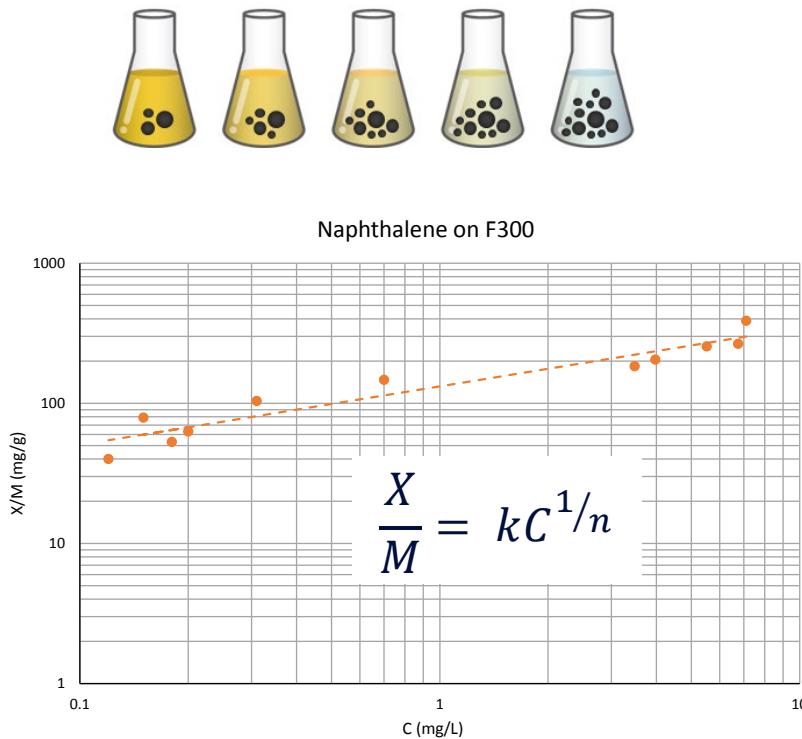
## Adsorbent

- Base Material
- Degree of Activation

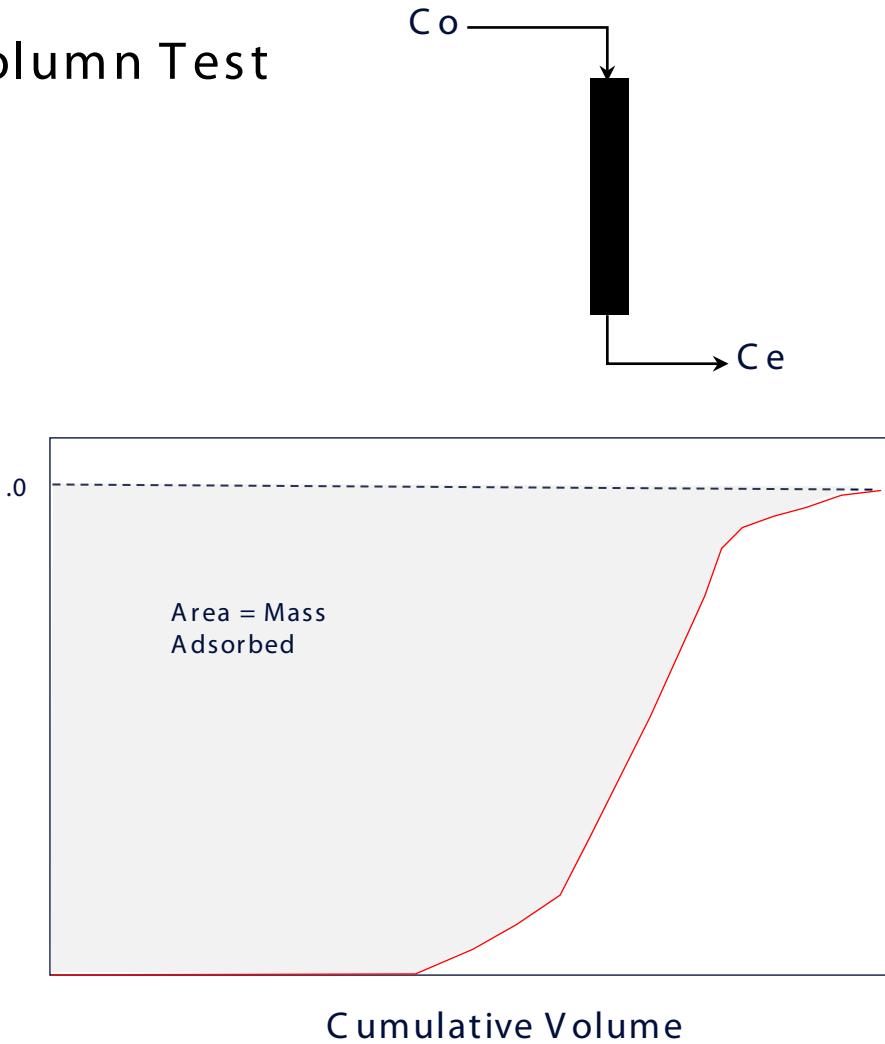


# Evaluating Equilibrium Loading Capacity

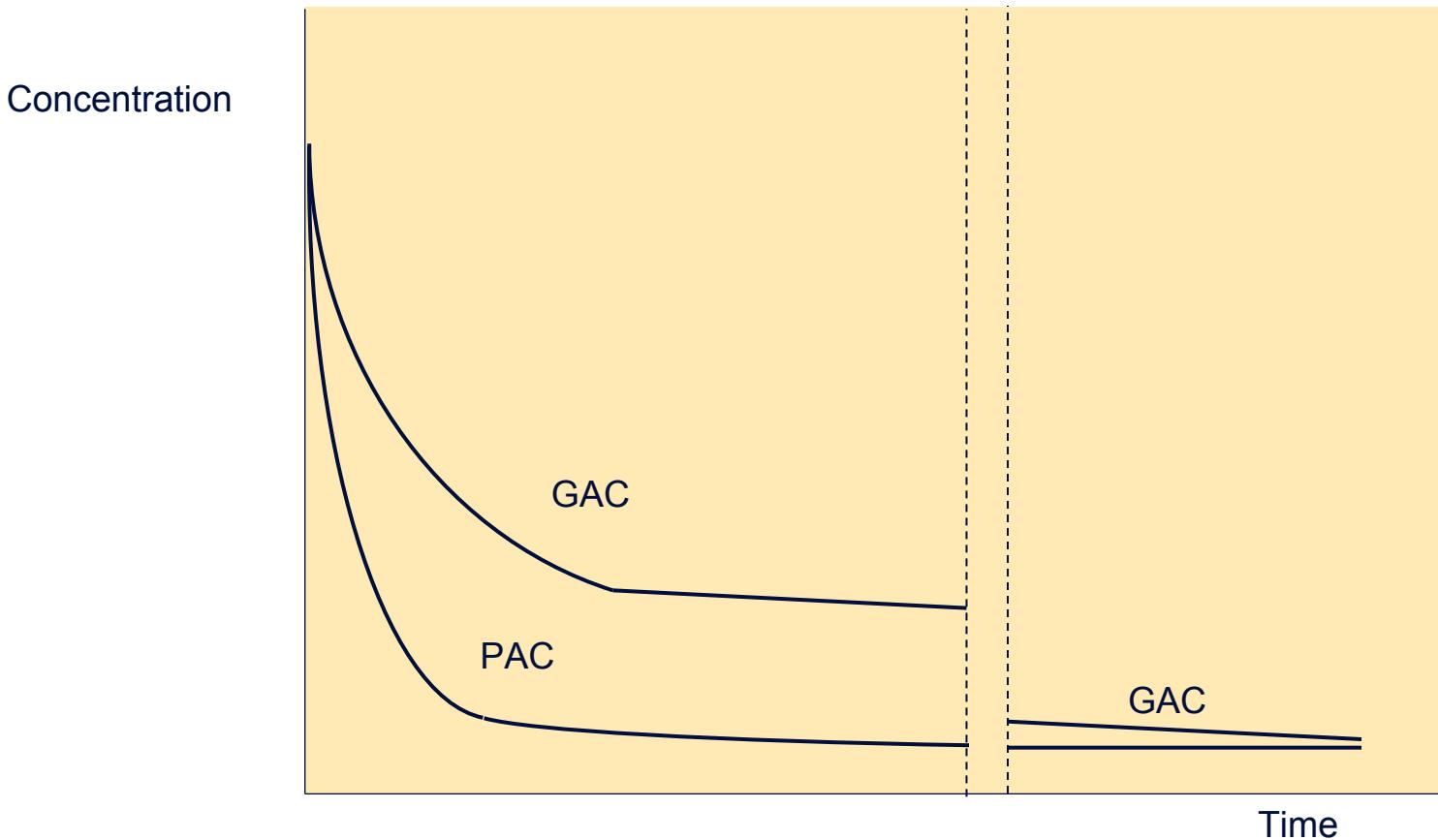
Isotherm (ASTM D 3860)



Column Test



# Evaluating Equilibrium Loading: Common Pitfall



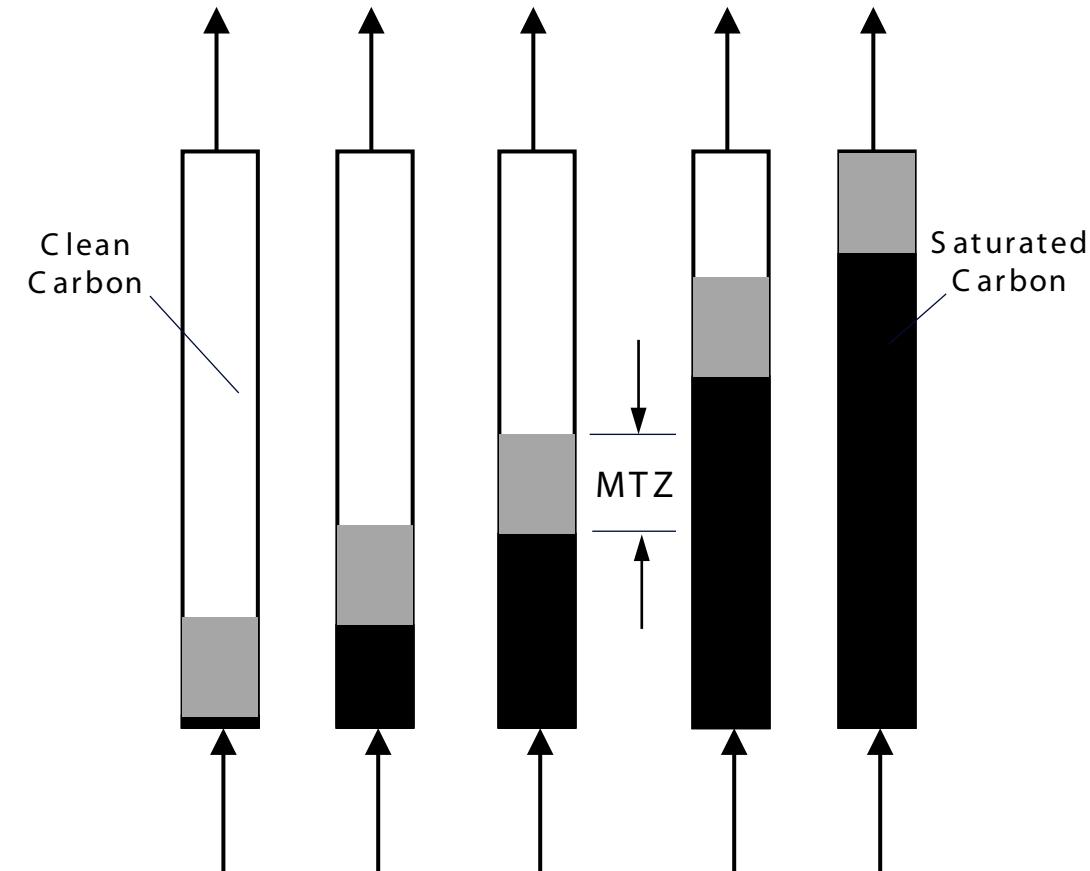
Peel et al. 1980; Randtke et al. 1983

- Not waiting long enough to approach equilibrium
- Not pulverizing the carbon (e.g., 95% -325 mesh)

# Kinetics

## Mass Transfer Zone (MTZ)

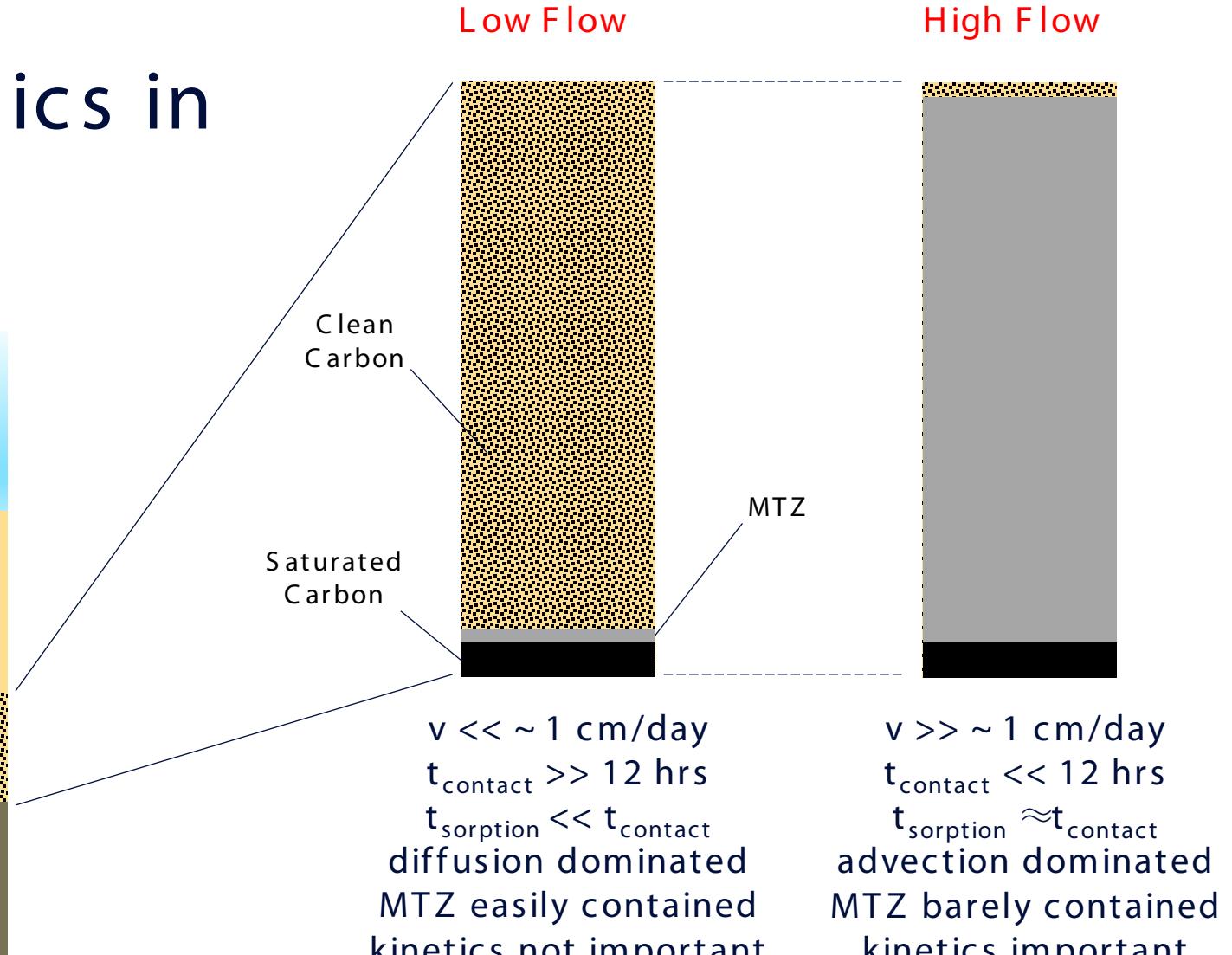
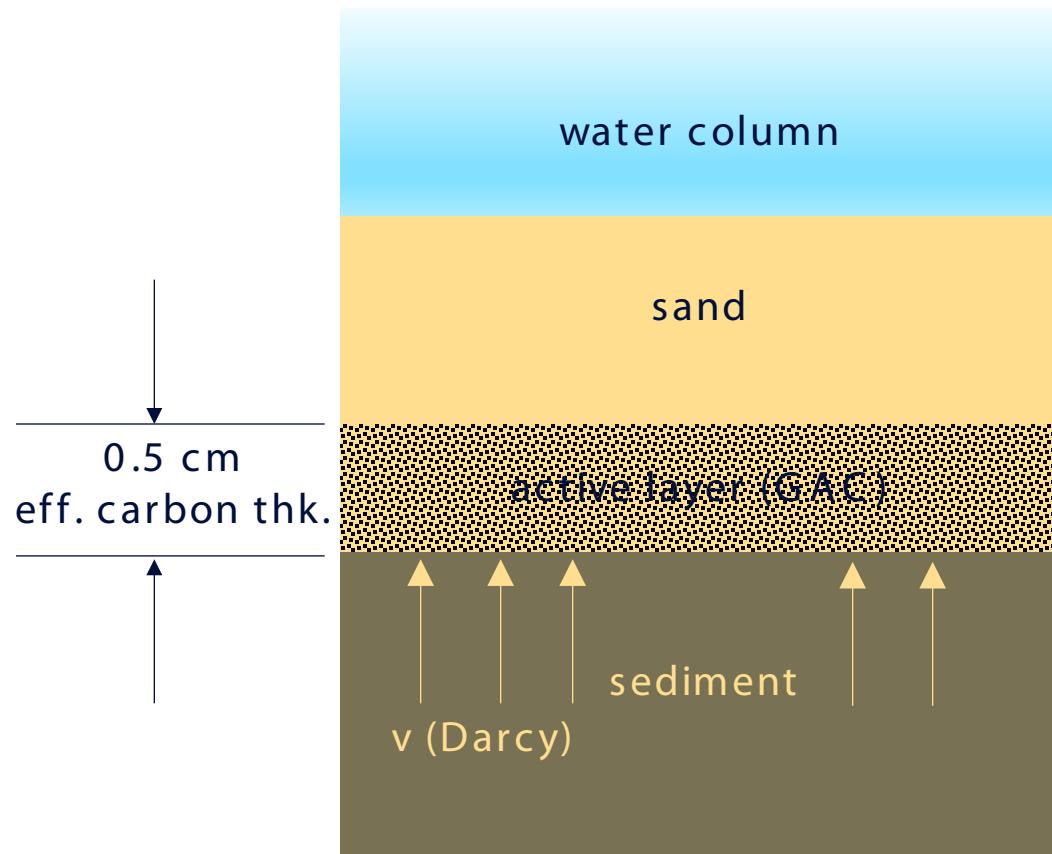
### Diffusion Processes



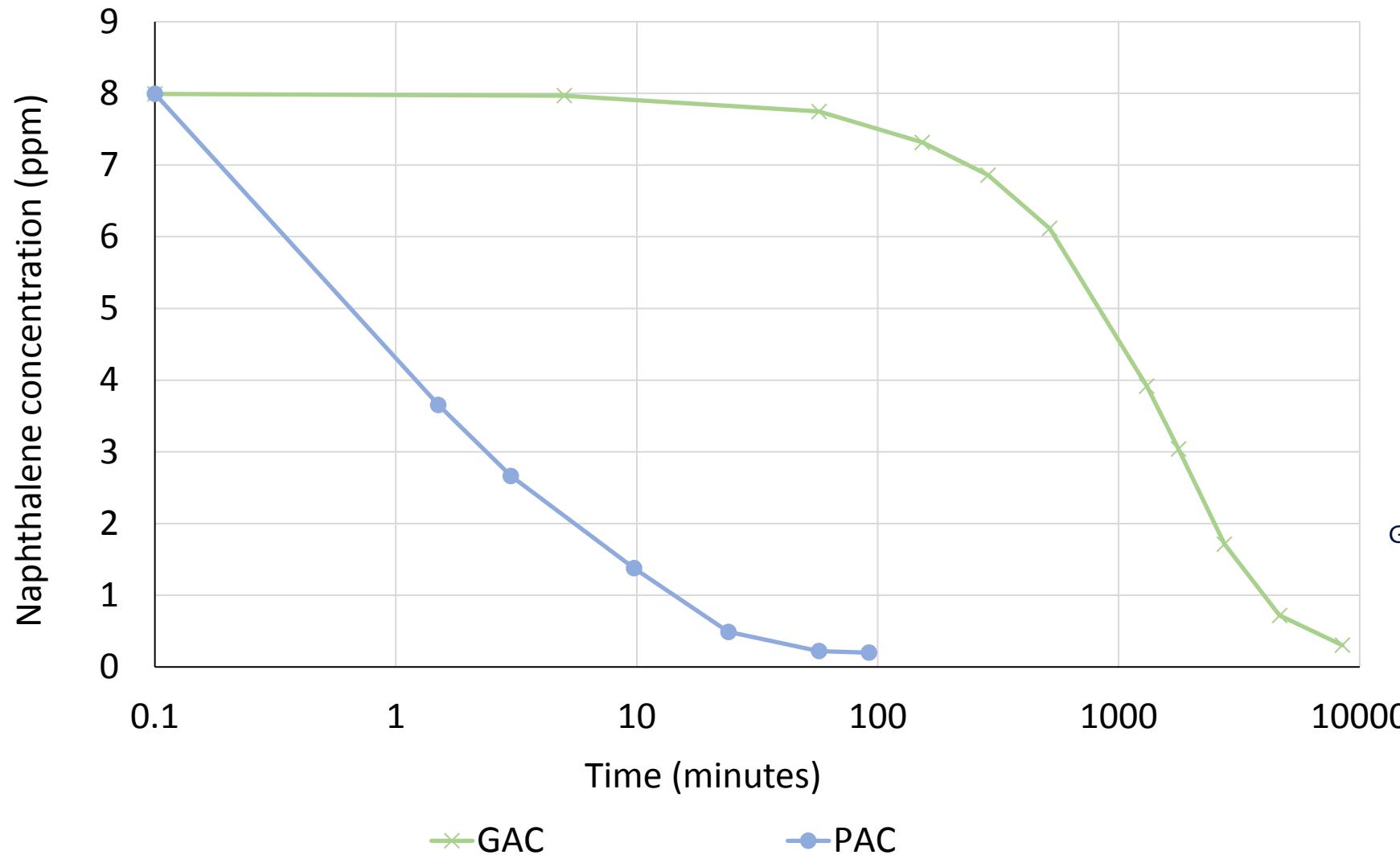
# Characteristics That Increase the Rate of Adsorption

- ↓ Carbon Particle Size (most important)
- ↑ Contaminant Diffusivity (Ficks's Law:  $J = -D \frac{dC}{dx}$ )
- ↑ Water Temperature ( ↓ Viscosity)
- ↑ Transport Pore Volume (between 1,000 and 100,000 Å, when bulk diffusion through large pores is limiting)

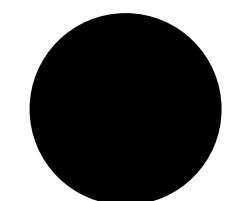
# Significance of Kinetics in Cap Designs



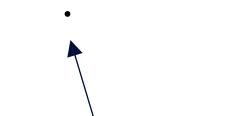
# Impact of Particle Size on Adsorption



Test Conditions:  
Adsorbate = naphthalene  
Solvent = distilled water  
 $C_0 = 8 \text{ mg/L}$   
Carbon dose = 50 mg/L  
Sample volume = 1000 mL  
Analysis: UV spectroscopy



GAC = F300 8x30



PAC = F300 sized to 95% -325 mesh

# Summary

- Equilibrium loading capacity is fundamental to all activated carbon applications
- Take care when evaluating capacity: particle size and contact time will impact results
- Adsorption kinetics may be significant for some applications, but not for all

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Thank you.